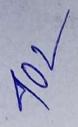
U.G. 5th Semester Examination-2024



CHEMISTRY

[HONOURS]

Discipline Specific Elective (DSE)

Course Code: CHEM-H-DSE-T-1A

(Polymer Chemistry)

Full Marks: 40

Time: $2\frac{1}{2}$ Hours

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

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1. Answer any five questions:

- $2 \times 5 = 10$
- a) Draw the structure of the polymers synthesized from
 - i) CH₂=CH-CO₂H and
 - ii) HO-(CH₂)₅-COOH
- Give an example of a co-polymer and a blockpolymer.
- c) Define η_J , η_{SP} and LVN.
- d) Why hydrogen halides are not suitable initiators for ionic polymerization?

- e) What is Zimm Plot?
- f) What is living polymerization?
- g) Write the Hildebrand equation and explain the terms.
- h) What is Carothel's equation?
- i) What are polyurethanes?
- j) Define Fluoro polymers.
- k) State lower and upper critical solution temperatures of polymer solution.
- 2. Answer any two questions: $5\times2=10$
 - a) What is polydispersity index? If 5 g of a monodisperse polystyrene sample of molecular weight 10,000 g mole⁻¹ is mixed with 15 g of another monodisperse polystyrene sample of molecular weight 20,000 g mole⁻¹, calculate the polydispersity of the polymer mixture. 1+4
 - b) Compare the essential features of the stepgrowth and chain-growth polymerization processes.
 - c) Define glass transition temperature (T_g) and state how to determine T_g.

- d) Compare the natures of different types of initiator and co-initiators used for ionic polymerization.
- 3. Answer any **two** questions: $10 \times 2 = 20$
 - a) What are the absolute and relative methods for determining the molecular weight of polymers? What is osmosis? Describé how M_n can be determined from osmotic pressure of a polymer solution.

 2+2+6
 - b) What is the difference between chain-growth and condensation polymerization? Explain why the DP_n value for step-growth polymerization changes slowly? Derive the kinetic rate expressions and DP_n for the catalyzed condensation reaction for a polyamide.

2+2+6

- c) What are radical initiators? How they are generated in situ? Describe the mechanism and kinetics of free radical polymerization of styrene.

 2+2+6
- d) Write notes on conducting polymers such as polyaniline, poly p-phenylene sulphide, polypyrrole, polythiophene.