

Notice

07.12.2024

All the Students of Semester V of Department of Chemistry are asked to submit the assignment given below on or before 17/12/2024.

SEM V (HONS)

Paper CC-12

1.Briefly discuss different types of molecular spectra.	4
2. Discuss the Born-Oppenheimer approximation	3
3. Briefly discuss excess pressure with example.	3

Paper: CC-11

- 1. The [CrF₆]³⁻ exhibits absorption bands at 14900, 22700 and 34400 cm⁻¹. Draw the orgel diagram and assign the electronic transitions. 3
- 2. Calculate the CFSE in terms of Do for the complexes $[Ni(H_2O)_6]^{+2}$ and $[NiCl_4]^{-2}$. Then comment on their stabilities. 3
- 3. Write a brief note on Jahn-Teller Distortion and show the splitting pattern of z-out and z-in distortion with brief explanation.

Paper: DSE 1A

1. What is meant by sustainability in green chemistry? Explain the twelve principles of green chemistry. Define 'atom economy' and 'E- Factor' in green chemistry. 1+2+1+1=5

2. What is the principle behind microwave-assisted reactions? Write down difference between microwave heating and traditional heating. Give examples of microwave assisted reaction in water solvent and organic solvent (give two examples in each case) 2+1+2 = 5

Paper DSE 1B

- 1. Write down the ingredients of cement. What is safety glass? Describe briefly the steps involved in the manufacture of glass with flow chart diagram. 1+1+3=5
- 2. What is battery? Write a short note on Li-ion battery. Write the cell reactions of lead storage battery. 1+2+2=5

SEM V (PROGRAM)

Analytical, Environmental and Industrial Chemistry Course code – CHEMGTDSE-1

- 1. What is the principle of thin-layer chromatography (TLC)?
- 2. Explain the formation of photochemical smog and its environmental effects.
- 3. Explain the concept of accuracy and precision in analytical results.
- 4. Explain the Haber process for ammonia production.

5. Explain the significance of Biochemical Oxygen Demand (BOD) and Chemical Oxygen Demand (COD) in water analysis.