■ 9153549620 ⊠dumkalcollege@gmail.com



DUMKAL COLLEGE

P.O-Basantapur,P.S-Dumkal,Dist.-Murshidabad,WestBengal, PIN-742406 (Govt.Aided, Affiliated to the: University of Kalyani Included under section 2(f) & 12 (B) of UGC Act.)

2nd CYCLE NAAC ACCREDITATION PROCESS-2024

CRITERIA: 1 – Curricular Aspects

Key Indicator: 1.3: Curriculum Enrichment

Metric: 1.3.2 -Supporting Document ***** Number of students undertaking project work/field work / internships during latest Completed academic year ***** Relevant Part of the Curriculum of different Subjects related to Project Work/Field Work/Dissertations etc

REPORT OF FIELD WORK 2022-2023

Title of the Report: SOCIAL AND ECONOMIC ASPECTS OF MANGALBARI CENSUS TOWN OF JALPAIGURI DISTRICT, WB

Prepared by: Semester V (Hons) Students of Department of Geography

No of Students: 51

Location of Fieldwork: Mangalbari Town

Dates of Fieldwork: 28 Oct-01 November. 2022

Supervisor/Guide: 1. Dr. Swati Mollah, Associate Prof. of Geography

2. Sri Arindom Biswas, Assistant Prof. of Geography

3. Dr. Prasenjit Kumar Mondal, Assistant Prof. of Geography

Attendance of Students

3 (1) 12 (1) 13 (1) 14 (1) 153 (1) 151 (1) 153 (1) 155 (1)	REG_NO 058224 058225 058226 058227 058228 058229 058230 058231 058232 058232	REG_YR 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022 2022	SIGNATURE Grandib Sarkar -Hafif Asod Mahdvo Hosson Manuar Hosson Manuar Hosson Manuar Sarkar Md. Abu Baukagi siddigi Mustakin Sk Ngbin Ghosh Ngbin Ghosh Ngbin Ghosh Nghu Bauka
3 (1) 12 (1) 13 (1) 14 (1) 153 (1) 151 (1) 153 (1) 155 (1)	058225 058226 058227 058228 058229 058230 058231 058232 058232	2022 2022 2022 2022 2022 2022 2022 202	-Hadij Asod Mahdin Hossan Manuar Hossan Manuar Hossain Manuar Sarkas Md. Abu Baukagi siddigi Mussalin Mussalin Mussalin Sk Ngbin Ghosh Najnun Sokir Sarkan
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55	058233	2022	1
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72	058234	2022	Robert Mandah
		2022	Mainell I aperatoria
76	058235	2022	Rimon SKa
77	058236	2022	Romio mondal
30	058237	2022	Sabir Ahmed
87	058238	2022	Sakil Ahmed
88	058239	2022	Sakil Ahmed
91	058240	2022	Samsul Molla
93	058241	2022	Sayan Day
94	058242	2022	Sheepberchip Source.
99	058243	2022	Souroav Mia
02	058244	2022	Sujoy Halder
06	058245	2022	Largy Kuman Saha
	 37 38 91 93 94 99 02 	87 058238 88 058239 91 058240 93 058241 94 058242 99 058243 02 058244 06 058245	87 058238 2022 88 058239 2022 91 058240 2022 93 058241 2022 94 058242 2022 99 058243 2022 99 058243 2022 99 058243 2022 05 058244 2022 02 058244 2022 06 058245 2022

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ANANYA PRAMANIK	56646	058187	2022	Ananga Anamanik
ARCHITA GHOSH	56648	058188	2022	Adichide Ghosh
ARPITA BISWAS	56650	058189	2022	Arpita Biswas
ΑSHIYA KHATUN	56652	058190	2022	Ashija Khatun
ΒΙΤΗΙ ΚΗΑΤΨΝ	56656	058191	2022	Brithe Khatun
DISHA MONDAL	56657	058192	2022	Disher Mondal
DURBA DAS	56658	058193	2022	Durba Das.
HASINA TUL KANIZ	56660	058194	2022	(Ab)
JINATAMAN BEGAM	56663	058195	2022	Jinatamon Begom
JUMANA KHATUN	56665	058196	2022	Jumana Khatun
NAJMUN FERDOUS	56671	058197	2022	Najmun Feredous
NEHA KHATUN	56673	058199	2022	Neha Khatun
NOWSIN NOWAR	56676	058200	2022	Nouisin Nowar
RAIMA AFJAL	56678	058201	2022	Raima Affal
RIMPA HALDER	56679	058202	2022	Rimpa Haldero
RIYA KHATUN	56680	058203	2022	Riya khatun
RIYAMONI SARKAR	56681	058204	2022	Rizamoni Sarkar
RUPA KHATUN ANSARY	56682	058205	2022	Rupa Khatun Ansang
SABNAM MUSTARI BANU	56683	058206	2022	(A6)
SAKILA KHATUN	56688	058209	2022	Sakila kha-lun
SOHANA SARMIN	56693	058210	2022	Sohana Sarmin.
SUMAIYA SULTANA	56695	058211	2022	Sumaiza Sultana
SUPRIYA GHOSH	56697	058212	2022	Supriza Ghosh
WAPNA BISWAS	56700	058213	2022	Sunpra Biswas
AMANNA KHATUN	56702	058215	2022	Tamanna Khastun
ANIA SULTANA	56703	058216	2022	Tanie Sultana
JMME HABIBA	56706	058217	2022	Umme Habiba
JMME SALAHA	56707	058218	2022	Umme Galacha
NJAN MONDAL	56714	058219	2022	ANJERMondal
ITTU BISWAS	56721	058221	2022	Rittle Bisway
ITTU BISWAS	56722	058222	2022	Bittle Biswaz
MDADUL SK	56724	058223	2022	Emdadul SK





Signature of Principal

• Principat Dumkal College -Basantapur, Murshidabad Signature of Teachers

oati Mollah

Ho D Department of Geography Dumkal College Basantapur, Murshidabad

REPORT OF PROJECT WORK 2022-2023

Title of the Report: PROBLEM OF FLOOD HAZARD IN JAMALPUR VILLAGE OF JALANGI BLOCK OF MURSHIDABAD DISTRICT, WB

2

Prepared by: Semester VI (Hons) Students of Department of Geography

No of Students: 51

Location of Fieldwork: Bamnabad Village

Period of Project: September 2023

Supervisor/Guide: Dr. Swati Mollah, Associate Prof. of Geography

Signature of Students Completing the Project Work

JINAIVIE	<u>-</u> .	-		
ANANYA PRAMANIK	56646	058187	2022	Ananga Anamanik
ARCHITA GHOSH	56648	058188	2022	Abichita Gchosh
ARPITA BISWAS	56650	058189	2022	Arpita Biswas
ASHIYA KHATUN	56652	058190	2022	Ashiya Khatun
ΒΙΤΗΙ ΚΗΑΤΨΝ	56656	058191	2022	Brithe Khatun
DISHA MONDAL	56657	058192	2022	Disha Mondal
DURBA DAS	56658	058193	2022	Durba Das.
HASINA TUL KANIZ	56660	058194	2022	(At)
JINATAMAN BEGAM	56663	058195	2022	Tinatamon Bego
JUMANA KHATUN	56665	058196	2022	Jumana Khatun
NAJMUN FERDOUS	56671	058197	2022	Najmun Ferodous
NEHA KHATUN	56673	058199	2022	Neha Khatun
NOWSIN NOWAR	56676	058200	2022	Nouisin Newar
RAIMA AFJAL	56678	058201	2022	Raima Affal
RIMPA HALDER	56679	058202	2022	Rimpa Halders
RIYA KHATUN	56680	058203	2022	Reya khatan
RIYAMONI SARKAR	56681	058204	2022	Rizamoni Sarkar
RUPA KHATUN ANSARY	56682	058205	2022	Rupa Knatun Ansamp
SABNAM MUSTARI BANU	56683	058206	2022	(A6)
SAKILA KHATUN	56688	058209	2022	Sakita kha-lun
OHANA SARMIN	56693	058210	2022	Schana Sarmin.
UMAIYA SULTANA	56695	058211	2022	Sumaizo Sultano
UPRIYA GHOSH	56697	058212	2022	suprita Ghosh
WAPNA BISWAS	56700	058213	2022	Swapna Biswas
AMANNA KHATUN	56702	058215	2022	Tamanna Khoutun
ANIA SULTANA	56703	058216	2022	Tanie Sultana
IMME HABIBA	56706	058217	2022	Umme Habiba
MME SALAHA	56707	058218	2022	Umme Galaha
NJAN MONDAL	56714	058219	2022	Antian mondai
ITTU BISWAS	56721	058221	2022	Ridde Bisway
ITTU BISWAS	56722	058222	2022	Bittle Bisucar
MDADUL SK	56724	058223	2022	Emdadul SK

REG_NO	REG_YR	NAME	Signature
052154	2021	TAJMIRA MONDAL	Tajmira Mondal
052155	2021	WASIFA FERDOWSI	Wasifa Ferdowsi
052156	2021	ABDUL GAFFAR MONDAL	Abdul Gaffar Monalel.
052157	2021	ABDULHALIM	Absent
052158	2021	AINUL SHAH	Aimed Shah
052159	2021	ALL MAMUN MANDAL	All Munmun Ma
052160	2021	ANOWAR HOSSAIN	Anowar Hossain_
052162	2021	BISWAJIT MANDAL	Biswoith mondal
081689	2021	DIBYENDU PRAMANIK	Dibyerdu Bramani
052164	2021	KANCHAN KUMAR PAL	Kanchan Kr. Pal.
052165	2021	KESHAB MONDAL	Keshab Mondal.
052166	2021	MD AKHIBUL ISLAM	Md. Anhibau Islam
052167	2021	MD NUHUNABI KHAN	Md Nuhunaki khom
052168	2021	NABAB SEKH	Nabab Sekh,
081691	2021	NAJIR HOSSAIN	Nay 12 Hossai h
052169	2021	NASIMUDDIN MOLLA	Nasimuddin Molla
052171	2021	NILAY MISTRI	Nilay Mistri
052172	2021	RAHUL MONDAL	Rahu Mondal
052173	2021	RASHIKUL MONDAL	Roshikul Mondal
052175	2021	RIJWAN SK	Rijwan SK
052176	2021	SABBIR AHAMMED	Sabbir Apamine
052179	2021	SAJID ALI	Sajid Ali
052180	2021	SAMIUL ISLAM	Sandr Dam
052181	2021	SAYAN MANDAL	Sagan Mandal
052182	2021	SOUBHIK SARKAR	Soublik Sarkaz
052183	2021	SOUMAK SAHA	Soumak Saha

Signature of Principal

2002.

. Principal Dumkal College Besantapur, Murshidabad Signature of Teachers

Locati Mollah Ho D Department of Geography Dumkai College Basantapur, Murshidabad



DUMKAL COLLEGE

DEPARTMENT OF CHEMISTRY Project Work Students List (2022-2023)

Sl. No	Programme Name	Programme CODE	Name	Session
1	Green approach for the synthesis of drugs	CHEMHTDSE-4	SATYAM DEBNATH	2022-23
2	Advancing catalytic converter technology for sustainable mobility	CHEMHTDSE-4	SAJJAD HOSSAIN	2022-23
3	Preparation, structure, bonding and reactions of ferrocene	CHEMHTDSE-4	MARUF AHAMED	2022-23
4	Advancing catalytic converter technology for sustainable mobility	CHEMHTDSE-4	ASIK SEKH	2022-23
5	Preparation, structure, bonding and reactions of ferrocene	CHEMHTDSE-4	NASIM ANJUM	2022-23

Principal Dumkal College Besantapur, Murshidabad



DEPARTMENT OF COMPUTER SCIENCE DUMKAL COLLEGE

B.Sc. Computer Science Hons. 6th Sem. Project information Course Code :UG-CMSH-DSE-PRO-604, Course Title :Project Work/Dissertation Session-2022-23

S i. N o	Name of the Student	Project Title	Project Guided by	Signature of Students	Remark
1	ABDUL HAI SIDDIKI ANSARY	Weather Web App	SADEKUL ISLAM	Abdul Hal side	
2	AYAN MUKHERJEE	TIC-TAC-TOE Game	SADEKUL ISLAM	Ayan Munimper	
3	AZIZ AL AMAN	Weather Web App	SADEKUL ISLAM	AZO'Z AL AMAM	
4	BULBUL AHAMED	CALCULATOR	SADEKUL ISLAM	Buchw Annund	
5	DEBASIS HAZRA	TIC-TAC-TOE Game	SADEKUL ISLAM	DebassisHazna	
6	SAKIL MOLLA	COOKIFY	SADEKUL ISLAM	SakilMolla	
7	SAYAN BARAL	CALCULATOR	SADEKUL ISLAM	Sayan Bazal	
8	SHANTANU GHOSH	CALCULATOR	SADEKUL ISLAM	Shantan-Ghosh.	
9	SUBRATA PRAMANIK	CALCULATOR	SADEKUL ISLAM	Subrata Prou	
1 0	TAMASOW ROY CHOWDHURY	CALCULATOR	SADEKUL ISLAM	Subrata Prou	

,Sadekar) Le lam,

Head of the Department Dept of computer Science, d Dumkal College, Murshidabad

DUMKAL COLLEGE

ENVIRONMENTAL STUDIES

TOPIC : STUDY OF RIVER ECOLOGY

Year	Name
2022-2023	ADORA KHATUN
2022-2023	AFSANA FARHA
2022-2023	AJIJA KHATUN
2022-2023	IMTISAN KHANAM MOMI
2022-2023	IRIN YEASMIN
2022-2023	JESMIN KHATUN
2022-2023	MEHEJAMIN SULTANA
2022-2023	NAJMA SULTANA PARVIN
2022-2023	NITASA GHOSH
2022-2023	RIYA KHATUN
2022-2023	SALMA KHATUN
2022-2023	SARMIM AKTAR ANSARY
2022-2023	TANIA SULTANA
2022-2023	ABU RAIHAN SEKH
2022-2023	ANIRUDDHA PAL
2022-2023	DILWAR HOSSAIN
2022-2023	HASIRUL ISLAM
2022-2023	MEHEBUB BADSHA
2022-2023	PARVEJ TOUFIK
2022-2023	RIAJ MONDAL AKASH
2022-2023	SAHID HASAN
2022-2023	SAMIM AKTAR KHAN
2022-2023	SARWAR ALAMIN SARKAR
2022-2023	SAWAN AKTAR MOLLAH
2022-2023	SOHON AKTAR
2022-2023	SUBHANKAR MONDAL
2022-2023	TOUSIF AHAMED SHAH
2022-2023	ASMAUL HOSNA
2022-2023	BIDISHA CHAKRABORTY
2022-2023	PROMA MONDAL
2022-2023	SAFIYA SULTANA
2022-2023	SAGNIK CHAKRABORTY
2022-2023	TUHIN AHAMED
2022-2023	HABIBUR RAHAMAN
2022-2023	ABIDA SULTANA
2022-2023	ANKITA PRAMANIK
2022-2023	AYESA KHATUN
2022-2023	AYESHA SIDDIKA
2022-2023	BABITA KHATUN
2022-2023	BARSHA KHATUN
2022-2023	DILBARA KHATUN

2022-2023	ELIJA KHATUN
2022-2023	HAFIJA KHATUN
2022-2023	HASINA KHATUN
2022-2023	
2022-2023	
2022-2023	
2022-2023	ISMATARA KHATUN
2022-2023	JANNATAN FERDOSA
2022-2023	JERIN AKTAR
2022-2023	KARUNA SARKAR
2022-2023	KOYEL MONDAL
2022-2023	MAFUJA KHATUN
2022-2023	MINAKSHI ROY
2022-2023	MISTU SINHA
2022-2023	MONALISA PARVEEN
2022-2023	NAZNIN SULTANA
2022-2023	NURTAJ KHATUN
2022-2023	PIU HALDER
2022-2023	PIYALI GHOSH
2022-2023	PUJA SAHA
2022-2023	PURNIMA KHATUN
2022-2023	SADIA KHATUN
2022-2023	SAMIMA KHATUN
2022-2023	SAMIMA NASRIN
2022-2023	SANTAHAR PARVIN
2022-2023	SOHELI SARMIN
2022-2023	SOHELI SULTANA
2022-2023	SOHELI SULTANA
2022-2023	SONIA KHATUN
2022-2023	SRABANI KHATUN
2022-2023	SUMAYA SULTANA
2022-2023	OYAHIDA RAHAMAN
2022-2023	TANHAJ SIRIN
2022-2023	TANIA KHATUN
2022-2023	TANIA SIDDIKA
2022-2023	TIASHA KAR
2022-2023	UMMA SALMA SABNAM
2022-2023	UMME HABIBA
2022-2023	ABDUL AZIZ
2022-2023	ABDUL HAMID
2022-2023	ABDUE HAMID ABDUS SALIM MANDAL
2022-2023	ADDOS SALIM MANDAL ADITYA MONDAL
2022-2023	AKTARUL ISLAM BISWAS
2022-2023	ALI ASGAR SK
2022-2023	
2022-2023	
2022-2023	BAPAN MONDAL

2022 2022	
2022-2023	BIBHAS SARKAR
2022-2023	
2022-2023	DAUD IBRAHIM HOSSAIN
2022-2023	DEBJIT MONDAL
2022-2023	GOPINATH SARKAR
2022-2023	HABIBUL ISLAM
2022-2023	IKBAL HOSSAIN
2022-2023	JASIM MONDAL
2022-2023	KABIRUL ISLAM
2022-2023	M M TOUFIQUE
2022-2023	MASADUL BISWAS
2022-2023	MASUD REJA
2022-2023	MASUM ALI REJA
2022-2023	MD ALAMIN ISLAM
2022-2023	MD MESBAUL RUIN BISWAS
2022-2023	MD NURABUL MONDAL
2022-2023	MD SARMAN ALI KHAN
2022-2023	MD SAYEED HASAN BISWAS
2022-2023	MOFACHHER HOSSAIN
2022-2023	MONIRUL SHAIKH
2022-2023	MOTIUR RAHAMAN
2022-2023	MUNTASIR MAMAN ANSARI
2022-2023	MURSALIM SK
2022-2023	MUSTAKIM SK
2022-2023	NAIMUR RAHAMAN
2022-2023	NAJIR SEKH
2022-2023	NAJIK SEKT
2022-2023	NAJNE ALAW BADSHA NAZRUL MONDAL
2022-2023	
2022-2023	
2022-2023	
2022-2023	OMOR FARUK
2022-2023	OSMAN JINNURRAIN
2022-2023	RAHUL ISLAM SHAH
2022-2023	RAIHAN KAYESH KHAN
2022-2023	RIAZ SK
2022-2023	RIMON SK
2022-2023	RINKU SHAIKH
2022-2023	ROCKY SARKAR
2022-2023	RONIT HALDER
2022-2023	SABIR MALITHA
2022-2023	SAHABUL MONDAL
2022-2023	SAID ALOM
2022-2023	SAIDUL ISLAM
2022-2023	SAJIB SHAIKH
2022-2023	SANKU MONDAL
2022-2023	SAYANDEEP SAHA

2022-2023 SELIM REJA 2022-2023 SIRAJUL ISLAM 2022-2023 SITAN SAHA 2022-2023 SOUMMYADIP ROY 2022-2023 SUBHAM MONDAL 2022-2023 SUBHAN KAR SHIL 2022-2023 SUBHANKAR SHIL 2022-2023 SUJP DAS 2022-2023 SUJAY HALDER 2022-2023 TANJIL MONDAL 2022-2023 TOUSIF AHAMED 2022-2023 MURSELIM MONDAL 2022-2023 HASINA KHATUN 2022-2023 JENIFA KHATUN 2022-2023 NAJMIN SULATAN SHAH 2022-2023 NASIMA NASRIN 2022-2023 NASIMA NASRIN 2022-2023 NASIMA AKTAR 2022-2023 NISAT AFRIN 2022-2023 NISAT AFRIN	
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2022-2023SOUMMYADIP ROY2022-2023SUBHAM MONDAL2022-2023SUBHANKAR SHIL2022-2023SUDIP DAS2022-2023SUJAY HALDER2022-2023TANJIL MONDAL2022-2023TOUSIF AHAMED2022-2023MURSELIM MONDAL2022-2023JENIFA KHATUN2022-2023NAJMIN SULATAN SHAH2022-2023NASIMA NASRIN2022-2023NASIMA AKTAR2022-2023NISAT AFRIN2022-2023NURE TAMANNA YASMIN	
2022-2023SUBHAM MONDAL2022-2023SUBHANKAR SHIL2022-2023SUDIP DAS2022-2023SUJAY HALDER2022-2023TANJIL MONDAL2022-2023TOUSIF AHAMED2022-2023MURSELIM MONDAL2022-2023HASINA KHATUN2022-2023JENIFA KHATUN2022-2023NAJMIN SULATAN SHAH2022-2023NASIMA NASRIN2022-2023NASIMA AKTAR2022-2023NISAT AFRIN2022-2023NISAT AFRIN	
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2022-2023	ASORA KHATUN
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2022-2023	AYRINA KHATUN
2022-2023	AZENARA KHATUN
2022-2023	AZIZA SULTANA
2022-2023	BABY NAJMIN
2022-2023	BABY SAHANA
2022-2023	BASIRA KHATUN
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2022-2023	ESMINA KHATUN
2022-2023	FARHANA KHATUN
2022-2023	FERDOSHA AKTAR BANU
2022-2023	HABIBA KHATUN
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2022-2023	JANNATON FERDOUS
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2022-2023	MINAKSHI KHATUN
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2022-2023	MOLLIKA KHATUN
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2022-2023	MONIKA KHATUN
2022-2023	MOSLIMA KHATUN
2022-2023	MOTIARA BEGAM
2022-2023	MOU DEBNATH
2022-2023	MST AZMIRA KHATUN
2022-2023	MST NAJIMA KHATUN
2022-2023	MUNSURA KHATUN
2022-2023	NAFISA JAHANA
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2022-2023	NASRIN AKTER BANU
2022-2023	NICE KHATUN
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2022-2023	NURJAHAN KHATUN
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2022-2023	PINJURA KHATUN
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	RACHITRA HALDER
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	SAHANAJ KHATUN
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	SAHEDINNECHHA KHATUN
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	SAHINA BANU
	SAHINA KHATUN
	GAJINA KHATUN
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	SAPIA PARVIN
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2022-2023	ABU HENA MONDAL
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2022-2023	ANEKUL MOLLA
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2022-2023	ANKAN SARKAR
2022-2023	ANOWAR HOSSAIN
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2022-2023	ANUAR HOSSAIN
2022-2023	ARIF MONDAL
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2022-2023	ASANUL ALAM
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2022-2023	ASIF IKBAL BISWAS
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2022-2023	ASIK SK
2022-2023	ASIM REJA
2022-2023	ASMAN ALI MONDAL
2022-2023	AYUB ALI SARDAR
2022-2023	BAHARAM MALITHA
2022-2023	BAPAN BISWAS
2022-2023	BAPON MOLLA

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2022-2023	BIKRAM DAS
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2022-2023	ELIAS ALI
2022-2023	EMAN SK
2022-2023	FARDIN AHAMMED
2022-2023	FARDIN AHAMMED ANSARY
2022-2023	FARIDUL ISLAM
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2022-2023	GOLAM AMBIA
2022-2023	GOLAM KIBRIA
2022-2023	GOLAM KIBRIA BISWAS
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2022-2023	GOLAM SAROWAR SEKH
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2022-2023	HADINUR SK
2022-2023	HAMIM HOSSAIN
2022-2023	HAMIM MONDAL
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2022-2023	HELAL MONDAL
2022-2023	HUMAYAN MONDAL
2022-2023	HUMAYUN MONDAL
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2022-2023	JAHID HASAN
2022-2023	JAHID HASAN BISWAS
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2022-2023	JASIM MONDAL
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2022-2023	JOYNAL ABEDIN
2022-2023	JUBAI AL MAMON MONDAL
2022-2023	JUEL HOSSAIN
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2022-2023	JUEL SAIKH
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2022-2023	JULIAS HOQUE
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2022-2023	JURAES SK
2022-2023	KAFI MOLLA
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2022-2023	KHALID HASAN ANSARY
2022-2023	KHOKAN PRAMANIK
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2022-2023	KIRAN KHAN
2022-2023	KOUSHIK DAS
2022-2023	LOKESH PRAMANIK
2022-2023	MAFIKUL ISLAM
2022-2023	MAHAIMIN HAQUE
2022-2023	MAHAMMAD ALI
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2022-2023	MANJARUL ISLAM
2022-2023	MANOWAR HOSSAIN MONDAL
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2022-2023	MASIDUL HASAN
2022-2023	MASRIKUL ANSARY
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2022-2023	MASUD ANSARY
2022-2023	MASUD KHAN
2022-2023	MASUM SK
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2022-2023	MD EBRAHIM MOLLA
2022-2023	MD HOSSAIN ALI
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2022-2023	ANOWAR HOSSAIN
2022-2023	MD MEHEBUB MALITHA
2022-2023	MD MUKLESUR RAHAMAN
2022-2023	MD NIJAMUDDIN MONDAL
2022-2023	MD RAFIKUL ANSARY
2022-2023	MD RAHIM MONDAL
2022-2023	MD RAHUL MONDAL
2022-2023	MD RAJIBUL ISLAM
2022-2023	MD RIAJUL MONDAL
2022-2023	MD SAGAR HOSSAIN
2022-2023	MD SAHARIYAR SK
2022-2023	MD SAID HALSANA
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2022-2023	MOHAIMIN REJA BISWAS
2022-2023	MONIRUL ISLAM SARDAR
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2022-2023	MUKLESHUR RAHAMAN
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2022-2023	RAJA MONDAL
2022-2023	RAJESH SHA
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2022-2023	RAJIB MOLLA
2022-2023	RAJIB PAL
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2022-2023	RAJU MONDAL
2022-2023	RAJU RANA
2022-2023	RAKESH SHEIKH
2022-2023	RAKIB HOSSEN
2022-2023	RAKIBUL MAHALDAR
2022-2023	RAKIBUL SHEIKH
2022-2023	RAKIBUL SK
2022-2023	RAMIZ REZA AHAMMED
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	RANJU SK
2022-2023	RASEL BISWAS
2022-2023	RASEL PARVEJ
2022-2023	RASEL RANA
2022-2023	RASHIDUL SK
2022-2023	RASHIKUL ANSARI
2022-2023	RASIDUL ISLAM
2022-2023	RASIDUL SK
2022-2023	REJANUR MALITHA
2022-2023	RIAJUL ISLAM
2022-2023	RIJUAN SARIF
2022-2023	RIPAN BHOWMICK
2022-2023	RIPON SARDAR
2022-2023	RIYAJ SAIKH
2022-2023	ROHET ALI MONDAL
2022-2023	ROKY MONDAL
2022-2023	RONI MANDAL
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2022-2023	SABIR AHAMMED MONDAL
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2022-2023	SAHIN REJA MONDAL
2022-2023	SAHIN SHAH
2022-2023	SAHIN SK
2022-2023	SAHINUR ALAM BISWAS
2022-2023	SAHINUR JAMAN
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2022-2023	SAIFIZ AHAMMED
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2022-2023	SAIFOLISLAWI
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2022-2023	SAKIL AHAMMED
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2022-2023	SAMIM HOSSAIN
2022-2023	SAMIM MANDAL
2022-2023	SAMIM SK
2022-2023	SAMIUL ISALM
2022-2023	SAMIUL MANDAL
2022-2023	SAMIUL MONDAL
2022-2023	SAMSUJJAMAN MONDAL
2022-2023	SAMSUL HUDA
2022-2023	SANOAR MONDAL
2022-2023	SANOWYAR MOLLA
2022-2023	SANUOR SK
2022-2023	SANUYAR HOSSAIN
2022-2023	SANUYAR SK
2022-2023	SARIOT MONDAL
2022-2023	SARTHAK SAHA
2022-2023	SARUARJAHAN MONDAL
2022-2023	SAYAN KARMAKAR
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2022-2023	SAYEED AKHTAR SK
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	SHRABON SARDAR
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2022-2023	SHUBHA DAS
2022-2023	SIBAJUL SK
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2022-2023	SIMARUL MONDAL
2022-2023	SIYAD DAFADAR
2022-2023	SOFIKUL ISLAM
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2022-2023	SOHEL RANA SK
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2022-2023	SOURAV ALI
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2022-2023	SOYEL SABBIR AHAMMED
2022-2023	SHARUKH HOSSAIN MONDAL
2022-2023	SUBHAJIT SARKAR
2022-2023	SUBRATA GHOSH
2022-2023	SUJAUDDIN MIA
2022-2023	SUKALPA PRAMANIK
2022-2023	SULTAN MONDAL
2022-2023	SULTAN SK
2022-2023	SUMAN DAS
2022-2023	SUMAN MALTHYA
2022-2023	SUMAN MONDAL
2022-2023	SUNARUDDIN ARINDA
2022-2023	SURAJ MONDAL
2022-2023	SUROJ AHAMMED MONDAL
2022-2023	SUROJ MIA
2022-2023	SURYA DAS
2022-2023	SUSHANTA MANDAL
2022-2023	SUVO MONDAL
2022-2023	SWAPAN SHEIKH
2022-2023	TAFIJUL SK
2022-2023	TAMANNUR BISWAS
2022-2023	TANBIR ALAM
2022-2023	TAUFIK OMAR
2022-2023	TOHID HOSSAIN
2022-2023	TOHIDUL MONDAL
2022-2023	TOHIDUL SK
2022-2023	TOWFIK AHAMMED
2022-2023	TOWSIF AHAMED
2022-2023	TUFAJJUL ISLAM
2022-2023	TUHIN MONDAL
2022-2023	TUHIN SK
2022-2023	UDAYADITYA SHIL
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2022-2023	ҮАКИВ SK
2022-2023	YEASIN SAIKH
2022-2023	YEASIN SEKH
2022-2023	YOUNUS ALI MONDAL
2022-2023	YUSUF ALI
2022-2023	ANKITA KUNDU
2022-2023	ARJINA KHATUN
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2022-2023	HOSNEWARA KHATUN
2022-2023	JHUMPA KHATUN
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2022-2023	ARIF MOLLA
2022-2023	ASIF MASUD
2022-2023	BASHIR AHMED
2022-2023	HASAN MONDAL
2022-2023	JOYNAL MONDAL
2022-2023	MD MINHAJ MURSHED KHAN
2022-2023	MD TANJIR HOSSAIN
2022-2023	NAYAN MONDAL
2022-2023	NIMESH PAL
2022-2023	RAHUL HALSANA
2022-2023	RAJESH MANDAL
2022-2023	SAFAIL MONDAL
2022-2023	SAHADUL MIA
2022-2023	SAHIN SARIAR
2022-2023	SAIBATUL HAMID
2022-2023	SAMIM ISLAM
2022-2023	SOUMEN MONDAL
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2022-2023	YEASIN ANSARY
2022-2023	FARUK AHMMED
2022-2023	LAAM MUNEME MIRZA GALIB
2022-2023	MAMUD SK
2022-2023	MD GOLAM MUSTAFA KHAN
2022-2023	MD SABIR SARKAR
2022-2023	MD SOFIK ALI
2022-2023	MOSAROF HOSSAIN MEHENA
2022-2023	NURTAJUL HOQUE
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2022-2023	ZEM BISWAS

TITLE: FIELD REPORT ON CASE STUDY OF ENVIRONMENTAL DISASTER IN BAMNABAD VILLAGE OF RANINAGAR II BLOCK OF MURSHIDABAD, WB

INTRODUCTION

Raninagar II is one of the 26 blocks of Murshidabad district of West Bengal. Our study area, Bamnabad Mouza is situated in the south extreme eastern part of Raninagar II block. Bamnabad is a village where the field survey has been done on river bank erosion problem. The study area is characterized by agrarian culture. Though the village is well connected by a local metalled road to State Highway 11 it is not well developed in terms of economic features. It has a number of educational institutions surrounding it. But it is a socially backward village having no awareness regarding health and education.

OBJECTIVES

These are the objectives of the field study:

- 1. To study the environmental characteristics of the study area
- 2. To study the river bank erosion problem and
- 3. To study the perception of the victims of the study area including preparedness

METHODOLOGY

In the pre-field study various official websites and published books are consulted for gathering the information regarding the study area. In the second stage the socio-economic data are collected. The study is based on primary data. The data are collected from field survey. For the presentation of the primary data appropriate diagrams are prepared.

STUDY AREA

Bamnabad mouza is located at 24°07′11″N 88°34′11″E in Raninagar II block. It is situated in the eastern part of the Raninagar II block. The total area of the village is 124 hectares. It is located in the right bank of River Padma. The study area is a rural area. The study area belongs to Rajapur Gram Panchayat.

Physical Characteristics of the Study Area

Relief and Physiography

The study area has elevation between 22 m from the sea level. It belongs to Padma-Bhagirathi-Bhairab Tract. The area is characterized by the recent to sub-recent geological age group. The lithological feature is that it consists of recent alluvium, clay, silt, sand, gravels etc.

Physio-graphically the Murshidabad is a part of lower Ganga Plain and the Bamnabad is situated in the Bagri physical sub-division. Bamnabad village has a unique geomorphic location lying in the bank of River Padma. It is situated in the low-lying part in the alluvial tract and is affected by the river bank erosion causing inundation during the rainy season. As per geological information in this region clay-type soil deposition has a great impact for its general geological formation and its physiography.



Drainage

The main river of the study area is River Padma. It flows in the eastern side of the study area. The river Bhagirathi and Bhairab flow in the western side of the village. Both the rivers are almost in decaying state due to heavy siltation. Both are spill channels of the River Ganga. In many places course the rivers appear as marshy land having no signs of Perennial River. As a result, in many places the river beds are used for paddy cultivation. There are a number of bills (wetlands) in the block. These blocks are products of paleo channels of Eastern Bengal Delta. But all the bills are now in drying stage. Waters are found in their beds only in the rainy season.

Climate

Murshidabad has a tropical wet-and-dry climate. The annual mean temperature is approximately 27 °C; monthly mean temperatures range from 17 °C to 35 °C (approximate figures). Summers are hot and humid with temperatures in the low 30's and during dry spells the maximum temperatures often exceed 40 °C during May and June. Winter tends to last for only about two and a half months, with seasonal lows dipping to 9 °C – 11 °C between December and January. On an average, May is the hottest month with daily average temperatures ranging from a low of 27 °C to a maximum of 40 °C, while January the coldest month has temperatures varying from a low of 12 °C to a maximum of 23 °C. Often during early summer, dusty squalls followed by spells of thunderstorm or hailstorms and heavy rains cum ice sleets lash the district, bringing relief from the humid heat. These thunderstorms are convective in nature, and is locally known as Kal baisakhi. Rains brought by the Bay of Bengal branch of South-West monsoon lash the city between June and September and supplies the district with most of its annual rainfall of approx 1,600 mm (62 in). The highest rainfall occurs during the monsoon in August approx 300 mm (12 in). Floods are common during Monsoon, causing loss of life, destruction of property, and loss of crops.

Soil

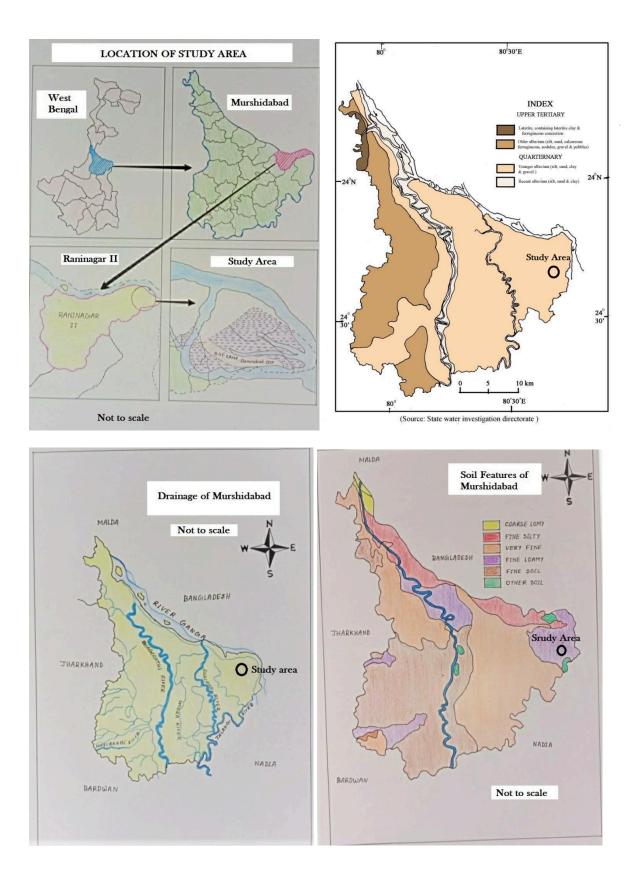
The study area is characterized by alluvial soil. The soil is formed from the recent alluvial brought by the rivers. The soil consists of very fine loams. So, the soil is very fertile for growing crops. Due to the very fertile land majority of the population depend on agriculture. The soil is exploited in its full extent. But now a day the fertility of soil has been degraded due to continuous use of fertilizers during farming. As a result, the soil becomes acidic.

Vegetation

Like the other part of the Murshidabad district the study area is characterized by deciduous trees. But due the settlement the vegetation cover is very low in the study area. Only orchards and few scattered trees are found in the study area. Among the prominent trees neem, mango, simul etc. are found in the area. Many of the households have kitchen gardens beside their house.

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Location map and Physical Characteristics of the Bamnabad village and its surrounding area



Socio-economic Characteristics of the Study area

Scoati Kollah

Ho D Department of Geography Dumkai College Basantapur, Murshidabae

Population Composition

The study area has total 733 households. The total population of the mouza is 4414 (2011). Among the total population the male and female population is almost same in numbers. So, the Sex Ratio in the study area is very good in terms of total number of female population per 1000 male population. The total number of populations below 6 years age is 11% of the total population. If we notice the female population under 6 years age, we find that the number is more than male population.

There is only 1% SC population live in the study area. The dominant religious category is Islam. Very few numbers of households follow Hinduism. The only language spoken in the village is Bangla.

Educational Level

The total percentage of literacy in the study area is 67%. It is quiet high with respect to the other villages. The rate of illiteracy is 33%. But the level of education is very low. Majority of literate population passed only primary level of education. There is a problem related with the availability of good school and educational facilities in the study area.

Infrastructural Facilities

The State Highway 11 connects the village to the nearest urban center, Raninagar. But the infrastructure is not developed in the study area. There are only 3 primary level school in the village. To go for Madhyamik and Higher Secondary school they have to go 5 to 10 km from their area. The Colleges are within 10 km from the villages. For getting public bus services the villagers have to go less than 5 kms and for railway services it is very far away, about 55 kms at Berhampore.

There are no good drinking water supply facilities in the villages. Only Arsenic Free water treatment plants were established few years back but due to non-maintenance of these plants they become defunct. There is no running water facility. The villagers use handpump for supply of drinking water.

Economic Status

The total number of working populations in the village is 923. So, working population percentage is very low. The main occupation of the workers is cultivation and agricultural labour. Many of them migrate to other states for earning. There is very few govt service holder in the study area.

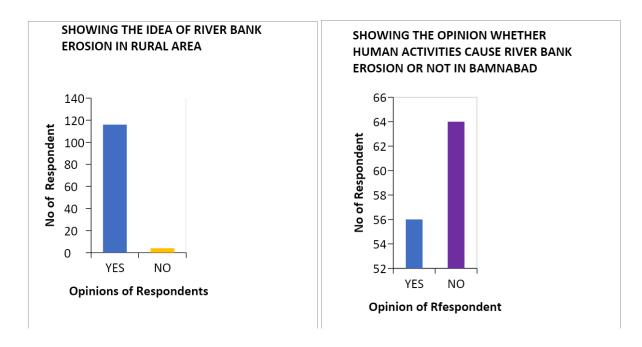
The total monthly income of the village is as low as Rs. 389840 and the total expenditure is Rs. 358830. This indicates the poor economic status of the study area.

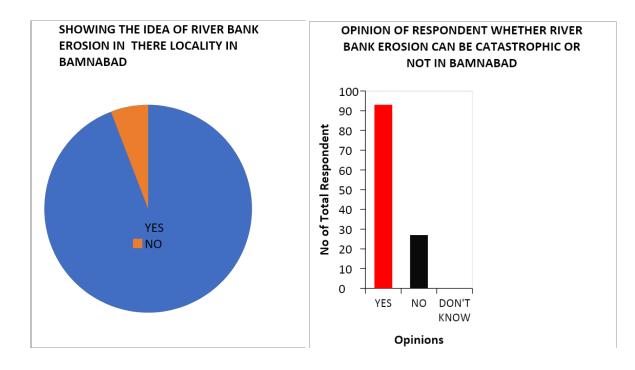
Land Use

The study area is characterized by two dominant land use that is settlement and cultivable land. Only 59 hectares land is under forest. More than 80% of the cultivated land is irrigated by groundwater. This causes over exploitation of groundwater in the study area. This also results arsenic pollution in the village.

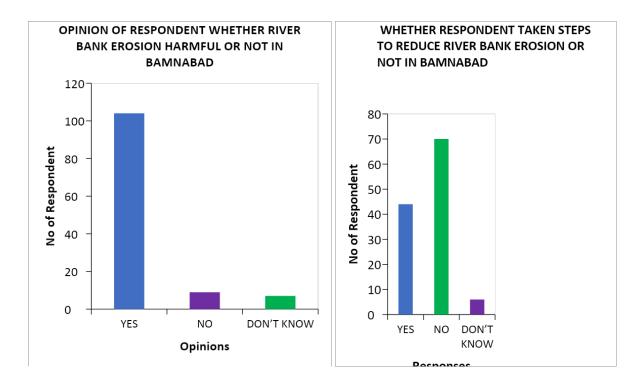
Diagrams showing the respondents perceptions, knowledge and preparedness level for river bank erosion in Bamnabad village

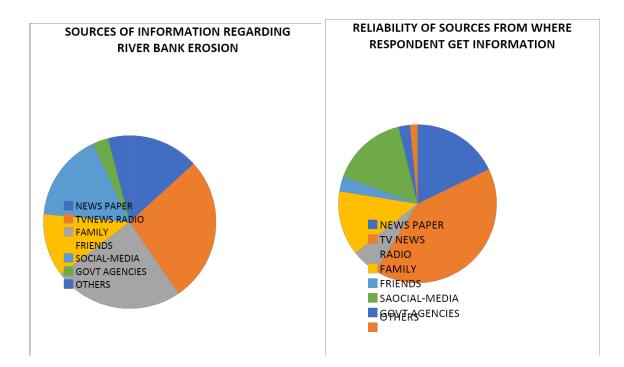




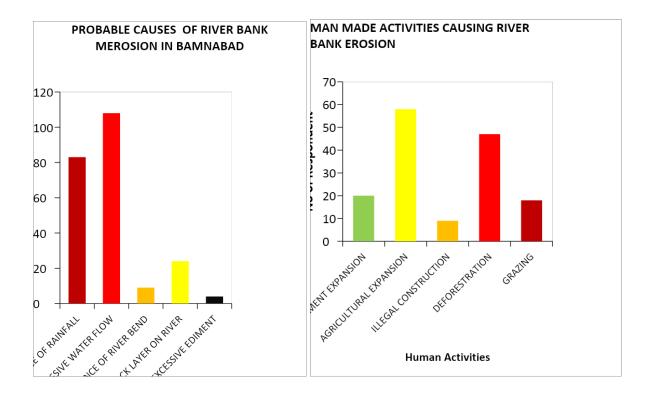


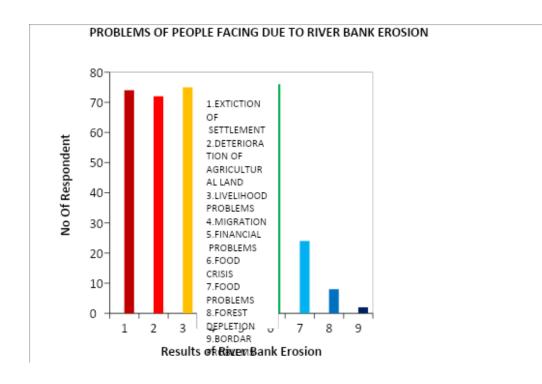
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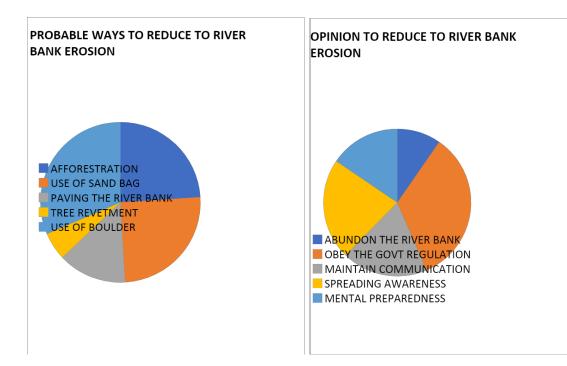


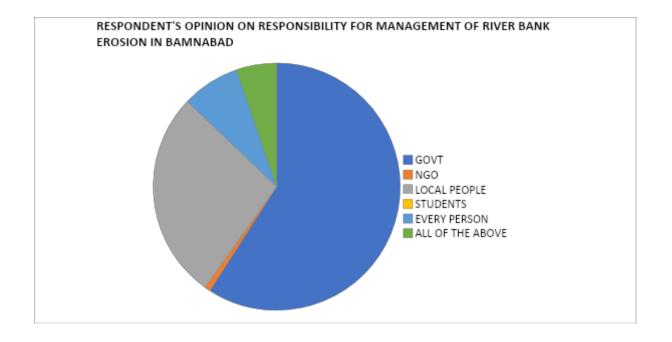
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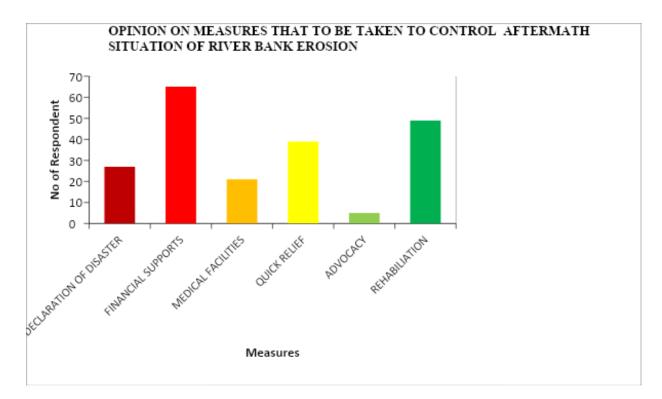


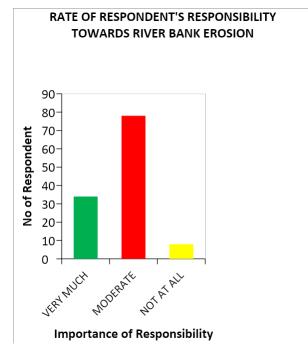
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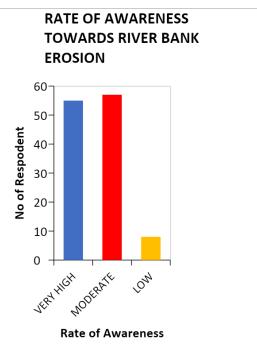




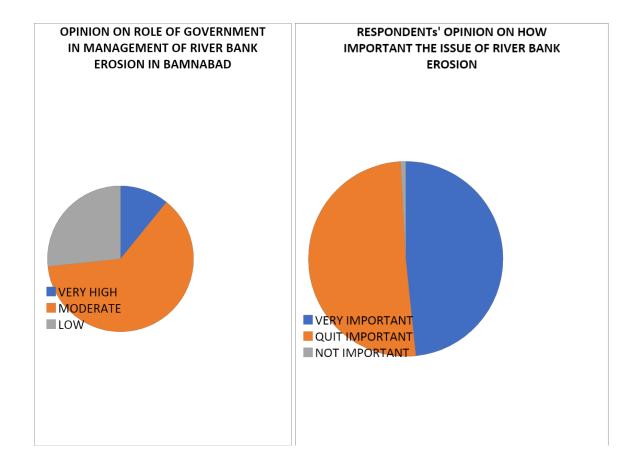
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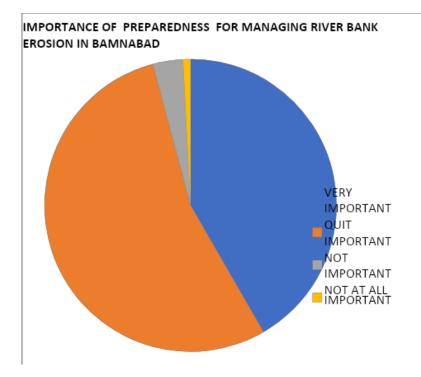






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Ho D Department of Geography Dumkal College Basantapur, Murshidabad

Field photos showing the physical, socio-economic features and river bank erosion problem in the study area



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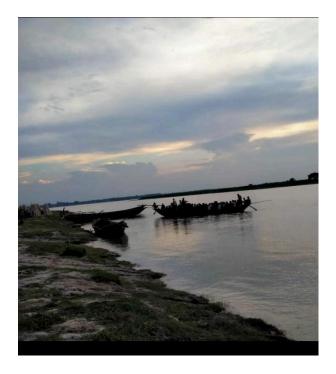




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CONCLUSION

The study area has a number of environmental problems. But the prominent problems are as follows-

- 1. River bank erosion
- 2. Inundation of settlement during rainy season
- 3. Arsenic pollution
- 4. Groundwater depletion

The villagers do not have much knowledge regarding disaster characteristics, preparedness, management etc. The lack of education and exposure are the main constraints here.

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UNIVERSITY OF KALYANI

Project report Submitted for the partial fulfillment of the B.sc degree

Dumkal College

Irin Aktar Soma

2022

Roll-2116213 No- 1949384 Regn No- 049373

Title of the project: CHLOROPHYLL AND HEMOGLOBIN

STRUCTURE AND FUNCTION

Name of the student:: Irin Aktar Soma,

We hereby forwarded this project Report for submission in partial fulfilment of the B.Sc. examination in Chemistry, 2022. We certify that the project work presented in this dissertation was carried out under our guidance.

Date:- 06-06-2022

Nabin ch. Maily

Saleha Khatun

Signature of the supervisor Signature of the supervisor

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CHLOROPHYLL AND HAEMOGLOBIN STRUCTURE AND FUNCTION

Introduction:

Chlorophylls are unique pigments with green colour and are found in diverse plants, algae, and cyanobacteria (Inanc, 2011). The term chlorophyll is derived from the Greek chloros meaning "green" and phyllon meaning "leaf". Isolation and naming of the chlorophyll were first carried out by Joseph Bienaimé Caventou (French pharmacist) and Pierre-Joseph Pelletier (French chemist) in 1817 (Gopiet al., 2014). Chlorophyll is made up of carbon and nitrogen atoms along with a magnesium ion in the central position. Chlorophyll is found in almost every green part of plants i.e. leaves and stem, within the chloroplast, the main organelle which contains the highest amount. Chloroplasts are found in the mesophyll layer, in the middle of plant leaves. Chloroplasts possess thylakoid membranes which contain green chlorophyll pigment. Chloroplast can be referred to as the "food factory" of the plant cell because it produces energy and glucose for the whole plant in association with CO₂, water, and sunlight. The name "Chlorophyll" was first given to the chloroplast of higher plants only, but later it was extended to all photosynthetic porphyrin pigments (Vernon and Seely, 1966). It comes under the special class of compounds called tetrapyrroles because it contains four pyrrole rings joined together with a covalent bond, as are vitamin B12 and the heme molecule (Willows, 2004).



Fig. 1. Leave containing green chlorophyll pigment

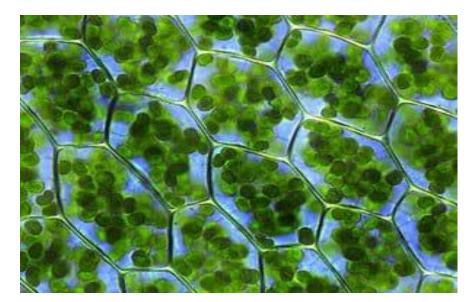
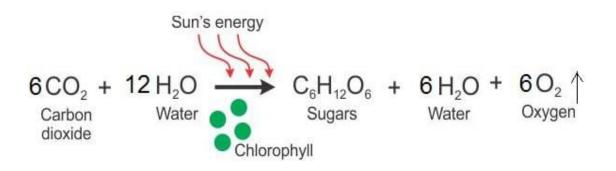


Fig. 2. Chlorophyll is responsible for the green colour of many plants and algae.

Chemistry of Chlorophylls:

The porphyrin unit has a very crucial role in nature because it participates in the fundamental skeleton of chlorophyll (Eugene and Govindjee, 1969). Many researchers have performed various experiments, and both analysed the chemical properties and elucidated the structure of chlorophyll molecules (Aronoff, 1966; eely, 1966). Research has revealed that the chlorophylls are tetrapyrroles, a cyclic form of porphyrin and chlorin (the parent molecule of all chlorophylls). This cyclic form creates an isocyclic ring with the help of -CH bridges. Chemically, chlorophyll possesses a magnesium ion in the central position which is found connected with the tetrapyrrole ring (Scheer, 1991). Moreover, chlorophylls are hydrophobic molecules because they contain phytol, an esterified isoprenoid C20 alcohol. The phytol (C20H30OH) possesses a double bond in the trans configuration (Gross, 1991).

Importance:



Scheme 1. Photosynthesis in chlorophyll

The main source of life on earth is the solar energy that is captured by green plants, algae, and various photosynthetic bacteria. Although there are different photosynthetic pigments such as carotenoids and phycobilins which entrap solar radiation, chlorophyll is the most important of these molecules. It converts solar energy into chemical energy that is used to build essential carbohydrate molecules (glucose) which are used as a food source for the whole plant (Hynninen and Leppakases, 2002). The process can be described by the equation Broadly, the conversion process of solar energy into chemical energy is called photosynthesis. Chlorophyll pigment absorbs blue light and red light of solar radiation at 430 nm and 660 nm, respectively, and it reflects the green spectrum (Inanc, 2011)

The light absorption of different photosynthetic pigments:

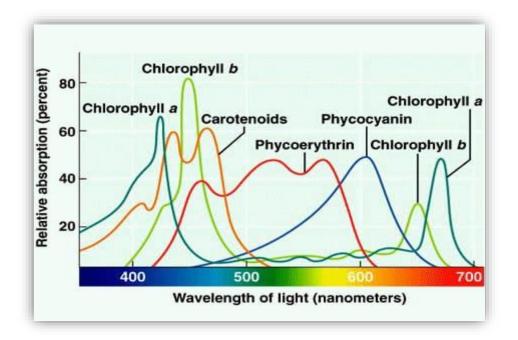


Fig. 4. Range of absorption by chlorophyll in the solar light spectrum.

Chloroplast:

The chlorophyll pigment is found in plant cells' chloroplasts. These chloroplasts act as a site of the photosynthesis process in both plants and blue-green algae.

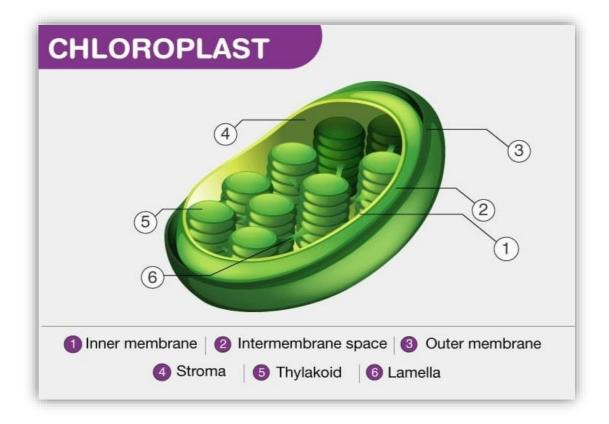


Fig. 5. Structure of chloroplast

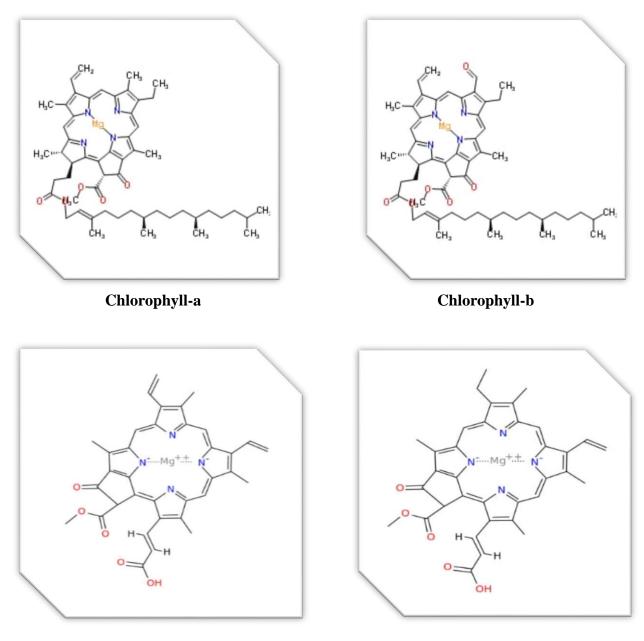
Usually, the chloroplasts align along the walls of the mesophyll. This helps them to receive optimum sunlight. The chloroplast has different membranes like grana, stroma, lamellae and thylakoids. The chlorophyll pigment is enclosed in the thylakoids of the chloroplast.

Types and Distribution of Chlorophylls:

The numbers of naturally occurring chlorophylls may not yet be fully known. Chlorophyll-a and chlorophyllb are the main components of photosystems in photosynthetic organisms. The types of different chlorophyll pigments present in nature are in decreasing order of importance. Initially, chlorophyll was classified into four – chlorophyll-a, chlorophyll-b, chlorophyll-c, and chlorophyll-d (Vernon and Seely, 1966) – but later a new type of chlorophyll was discovered within stromatolite (a hard rock structure made by cyanobacteria) in western Australia, which was named chlorophyll-f. Thus eventually chlorophyll was divided into five classes, a, b, c, d, and f.

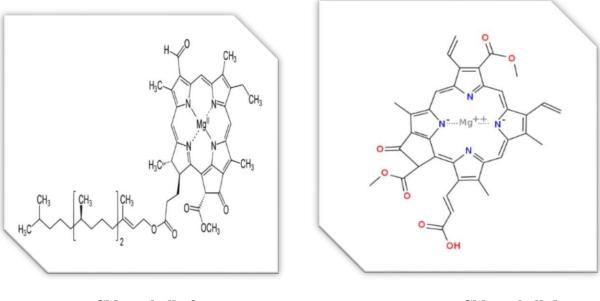
Structures of Different Chlorophylls:

Chemically, the basic skeleton of chlorophyll is composed of a cyclic tetrapyrrole ring, which is a large planar structure of asymmetric arrangement in which the four pyrrole rings are joined together by methine (-CH=) bridges and four nitrogen atoms are coordinated with a central metal atom. In addition, they have a phytol group that confers a hydrophobic characteristic; the metal-bound to the chlorophylls is magnesium. Therefore, this structure contains a chromophore of several conjugated double bonds responsible for absorbing light in the visible region, i.e. red (peak at 670-680 nm) and blue (peak at 435-455 nm). The reflection and/or transmission of the non-absorbed green light (intermediate wave-length) give the characteristic green colour to plants and chlorophyll solutions. Higher plants, ferns, mosses, green algae, and the prokaryotic organism prochloron only have two chlorophylls (a and b); the remaining chlorophylls are present in algae and bacteria. The structures of different chlorophylls are described below.



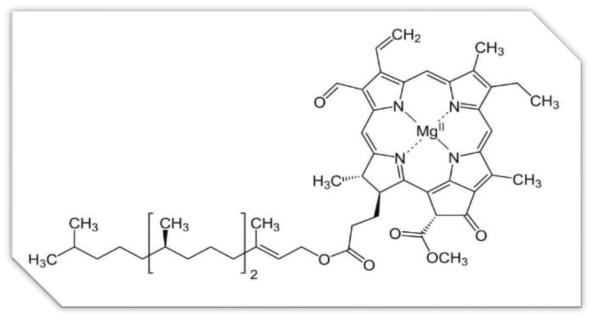


Chlorophyll-c1



Chlorophyll-c3

Chlorophyll-d



Chlorophyll-f

Fig. 6. Structure of various types of chlorophyll

Biological Function of Chlorophyll :

Plants use both forms of chlorophyll to collect energy from light. Chlorophyll is concentrated in the thylakoid membranes of chloroplasts. Chloroplasts are the organelles in which photosynthesis takes place. The thylakoids are small sacs of the membrane, stacked on top of each other. Embedded in these membranes are a variety of proteins that surround chlorophyll. These proteins work together to transfer the energy from light, through chlorophyll, and into the bonds of ATP – the energy transferring molecule of cells. ATP can then be used in the calvin cycle or dark cycle to create sugars.

The series of proteins that transfer energy from light and channel it into the synthesis of sugars are known as photosystems. The entire process, both light and dark cycles together, is known as photosynthesis and occurs

in plants, algae, and some bacteria. These organisms take in carbon dioxide (CO_2), water (H_2O) and sunlight to produce glucose. They can use this glucose in the process of cellular respiration to create ATP or they can combine the glucose into more complex molecules to be stored.

The function of photosynthesis:

The chlorophyll acts as a chromophore in photosynthesis. The extensive Conjugation in the chlorin ring allows the electron transition π (HOMO) $\rightarrow \pi^*$ (LUMO) in the visible region. One absorption band arises in the region 430 - 480 nm (i.e. blue light) white the other band appears in the region 645-680 nm (i.e. red light). Chlorophyll looks green because it absorbs red and blue light. These light-harvesting pigments act as a molecular antenna to absorb the solar energy which is transferred to a centre where a chemical reaction goes on. Thus chlorophylls present in the reaction centre convert solar energy to chemical energy.

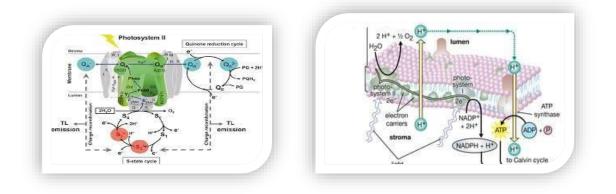


Fig. 7. Schematic diagram of photosynthesis

Chlorophyll catalyses the reduction of NADP⁺(to NADPH) and oxidation of H₂O (to O₂) in the presence of light. The electron flowers from H_2 too NADP⁺ through an electron transport chain (p-680 - p-700) which is known as Zschemem. The whole process is divided into photosystem-I and photosystem-II.

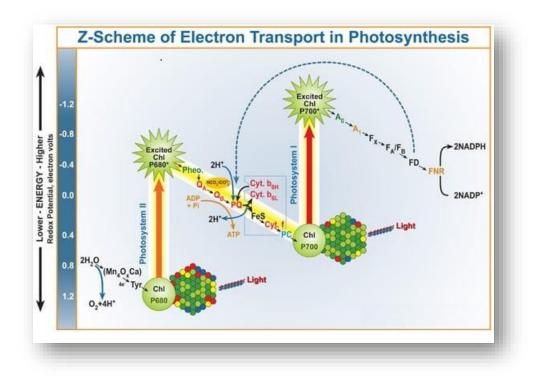


Fig. 8. Electron transport during photosynthesis

Photosystem-I (abbreviated as ps-I or p-700, stands for pigment) which is excited by the wavelength of light in the region 700 nm (or lower) generates a strong reductant to bring about the reaction of NADP⁺ to NADPH. Photosystem-II (ps-II or P-680) was the light of wavelength 680 nm or lower to produce a very strong oxidant to oxidise H₂O to O₂. PS-II uses chlorophyll-a, while ps-II was chlorophyll-a². When chlorophyll (present in ps-I) is excited it can act both as a better reducing agent and also a better oxidising agent. Thus the exciting chlorophylls can initiate a series of redox reactions. The electron transport chain from pheophytin to chl-a₁. The carriers are arranged in the increasing order of the reduction potential.

Chlorophyll in the production of oxygen:

A by-product of photosynthesis is oxygen. Plants can use this oxygen in cellular respiration, but they also release excess oxygen into the air. This oxygen helps in supporting life on Earth. The oxygen is produced in the first part of the light cycle of photosynthesis. Plants split water molecules to produce electrons, hydrogen ions, and diatomic oxygen (O₂). The electrons supply the electron transport chain that drives ATP production. The oxygen is released into the air. In this way, all the oxygen we breathe is produced.

Benefits of Chlorophyll:

Because of chlorophyll, all life on Earth is possible. The first benefit of chlorophyll is sugar, produced through the process of ATP which is driven by chlorophyll. Plants, as primary producers, produce the basis

of the food chain. All other organisms in the food chain rely on the sugars plants create to sustain life. While the top predators in a food chain may never eat a single plant, they most certainly eat herbivores. These herbivores only eat plants, and grow and create muscle by digesting and utilizing plant nutrients. The accumulation of these nutrients in nature would not be possible without chlorophyll. The second benefit realized by all organisms is oxygen. While chlorophyll does not produce oxygen directly, chlorophyll and the complex of proteins it is associated with the transfer of electrons to molecules like ATP and NADPH, which can hold energy in bonds. The need for electrons to drive this process causes water molecules to be split, creating oxygen. This oxygen is released into the atmosphere. Plants, algae, and cyanobacteria, produce all of the oxygen in the atmosphere. All other animals, and most plants, need this oxygen to survive.

HAEMOGLOBIN STRUCTURE AND FUNCTION

Introduction:

Haemoglobin (Hb) is the most studied of the heme-containing globulin proteins and yet is not fully understood. It was one of the first proteins to be studied by X-ray crystallography and earned Max Perutz the Nobel Prize in Chemistry in 1962. The structural studies provided a plethora of data that offered glimpses of the magnificent molecular mechanisms behind Hb physiological functions. Haemoglobin is a polyfunctional molecule that is involved in several functions, such as catalytic (nitrite reductase, NO dioxygenase, monooxygenase, alkyl hydro peroxidase, esterase, lipoxygenase); nitric oxide metabolism; metabolic reprogramming; pH regulation and maintaining redox balance (Kosmachevskaya and Topunov 2018). This chapter, however, focuses on Hb primary function of oxygen transport and how mutations, endogenous or exogenous ligands or effectors affect Hb allostery.

Chemistry of Haemoglobin:

This molecule represents the most populous faction of the bloodstream proteins. About 97% of the dry weight of red cells is all haemoglobin, and the average human circulatory system holds about 750g of this stuff (compare that to, for example, albumin, of which there is only 200g). You really could just describe the circulatory system as a large branching organ filled mainly with haemoglobin. The protein itself is about 64 kDa and dissolves very well in water (for example, the contents of a red cell is an almost completely saturated solution of haemoglobin, at around 330g/L).

It is not an especially stable molecule, or rather it performs poorly when stress-tested in conditions that lay outside of the normal physiological norms. It begins to denature if heated to 50° C, and by 65° C most of it

becomes irreversibly destroyed (Rieder et al., 1970). If you drop the pH to 3.0 or below, it unfolds and the haem falls off. In case you want to play with some at home, powdered bovine haemoglobin is available (it has a good shelf life, and you reconstitute it as you want). It is used to enrich agar, create calibration solutions for measurement instruments, and create culinary abominations. Before you ask, yes acellular free haemoglobin solutions have also been considered as a resuscitation fluid, but unfortunately it is hideously toxic (more on that later).

Structure of haemoglobin:

Haemoglobin is a tetramer composed of four subunits, two α and two β . These subunits form two identical $\alpha\beta$ dimers. Each subunit has a haem group. The quaternary structure of haemoglobin is essential for positive cooperativity. Each time one of the monomers binds oxygen, the molecule of haemoglobin undergoes a conformational change. This changes the equilibrium constant for the next O₂ molecule to bind the next subunit (increasing the affinity for O₂) This is described as a transition from the T (Tense) deoxygenated state to the R (Relaxed) oxygenated state, though a series of intermediate T-like and R-like states. This property is responsible for the sigmoid shape of the oxygen-haemoglobin dissociation curve.

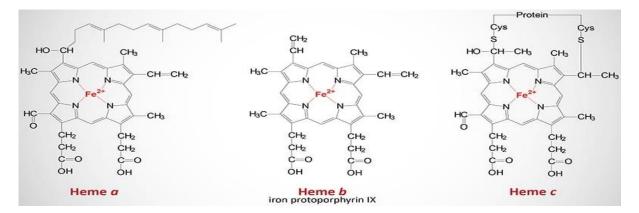


Fig. 9. Structure of various forms of haem

Haem groups are iron-containing molecules of protoporphyrin-IX with physiologically essential properties: Hydrophilic cores and hydrophobic external chains, for positive cooperativity. Able to bind different gases (O_2 , NO, CO, H₂S etc). Switching between Fe³⁺ and Fe²⁺ states allows it to participate in oxidation-reduction reactions.

The function of haemoglobin:

Oxygen transport:

Haemoglobin increases the oxygen-carrying capacity of blood by 50-100 times. Its affinity for oxygen is described by the oxygen-haemoglobin dissociation curve, which has a sigmoid shape. Affinity for O_2 is

increased under conditions of high PO₂, increasing its ability to collect oxygen from the lungs. Affinity for O_2 is decreased under conditions of low PO₂, enhancing the unloading of oxygen in the tissues.

Carbon dioxide transport:

Haemoglobin binds CO_2 with a high affinity in its deoxygenated state (the Haldane effect), and releases it when it becomes oxygenated and loses its affinity for CO_2 (the reverse Haldane effect). About 10-20% of the total carriage of CO_2 in the blood is owing to this mechanism.

Buffering:

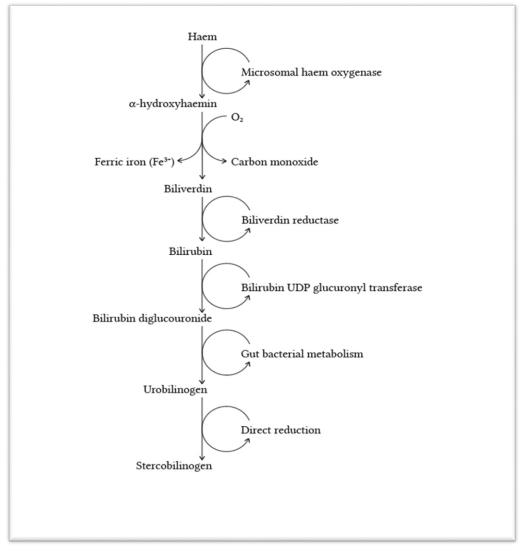
Haemoglobin has many histidine residue groups with a protonation pKa close to 6.8. This allows it to act as a buffer at physiological pH, as these histidine molecules act as proton acceptors. This buffering capacity increases in its deoxygenated state. As haemoglobin is one of the most populous proteins, it contributes a large proportion of the total buffering in the extracellular fluid, and this is incorporated into the calculation of the standard base excess. The total buffering power of haemoglobin is thought to account for up to 50-60% of the total buffering capacity of the blood, the rest being attributable to the bicarbonate and other proteins.

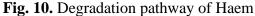
Nitric oxide scavenging:

Nitric oxide binds to the ferrous (Fe²⁺) iron with great affinity, creating nitrosyl haemoglobin. This reaction is rate-limited by the fact that haemoglobin is packaged in erythrocytes, which increases the useful half-life of nitric oxide. By removing nitric oxide from the peripheral circulation, haemoglobin contributes to the regional autoregulation of blood flow This mechanism is the most physiologically important mechanism for limiting nitric oxide bioactivity Clinical implications and examples of this include Hypoxic pulmonary vasoconstriction and pulmonary hypertension seen with polycythaemia (Deem, 2004). Sickle cell vasoconstrictive crisis (Mack and Kato, 2004)

Metabolic fate of haem:

The lysosomal breakdown of haemoglobin monomers releases the haem molecule. Right there in the macrophage, it is degraded into its constituent parts, liberating iron for reuse by the organism. One possible way of presenting this pathway is by a flow diagram.





Role of Haemoglobin (Hb) and Myoglobin (Mb) in Biological Systems:

Oxygen is inhaled into the lung. In human lungs, partial pressure of O_2 is high, the inhaled O_2 binds Hb present in blood in the lungs to form oxygenated haemoglobin (HbO₂), called oxyhemoglobin. We know that in Hb the sixth coordination position of Fe(II) is probably occupied by an H2O molecule which is trans to the histidine N-atom. When Hb binds with O2 to form a molecule present in Hb is reversibly replaced by the O_2 molecule to form HbO2.



Hb (containing coordinated H2O molecule) + O2 ==== HbO2 +H2O

As HbO2 runs to the muscular tissues through arteries, the partial pressure of O2 in muscle tissues decreases and HbO2 liberates O2.

HbO₂
$$\longrightarrow$$
 Hb + O₂

The liberated free O2 is taken up by myoglobin (Mb) to form oxygenated myoglobin (MbO2). This is called oxygenation of Mb.

Oxygenation of Mb

 $Mb + O_2 -$

As the blood runs through the arteries to the tissues, O2 bound with MbO2 is set free. This free O2 is used in the combustion (oxidation) of food (glucose, C6H12O6).

C6H12O6 + O2 6 CO2 + 6 H2O + 38 ATP (Energy)

The oxidation of glucose to carbon dioxide is an energy-releasing process in which energy in the form of ATP is produced. This energy is utilised by living organisms to perform their various metabolic activities, and for maintaining their body temperature. Water produced in the above reaction is retained in the body while CO_2 is exhaled out and haemoglobin goes to the lungs for reuse. The above discussion makes it evident that the function of haemoglobin in our body tissue supplies oxygen to various parts of the body. Thus haemoglobin is an oxygen carrier.

The Bohr effect:

The Bohr effect is a phenomenon discovered by Danish physiologist Christian Bohr in 1904. Haemoglobin's oxygen binding affinity is inversely related both to the pH and the concentration of carbon dioxide. Since carbon dioxide reacts with water to form carbonic acid, an increase in CO₂ results in a decrease in blood pH, resulting in haemoglobin protein releasing their load of oxygen. Conversely, a decrease in carbon dioxide provokes an increase in pH, which results in haemoglobin picking up more oxygen. The Bohr effect increases the efficiency of the oxygen transport through the blood.

References:

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- Thomas, Caroline, and Andrew B. Lumb. "Physiology of haemoglobin." Continuing Education in Anaesthesia, Critical Care & Pain 12.5 (2012): 251-256.
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Weather Web App

An Integrated Web Application Utilizing Real-Time Weather API Data to Deliver Comprehensive Forecasts, Climate Trends

> by Aziz Al Aman & Abdul Hai Siddiki Ansary

Iceland

Weather in Iceland

Q

6.13°C

Overcast Clouds

Humidity: 83% Wind speed: 1.58 km/h

Weather Web App: Learn how to build a weather web app using JS, CSS, HTML, and the OpenWeatherMap API. This presentation will cover everything from setting up your environment to making API calls and displaying the weather data on your web page.



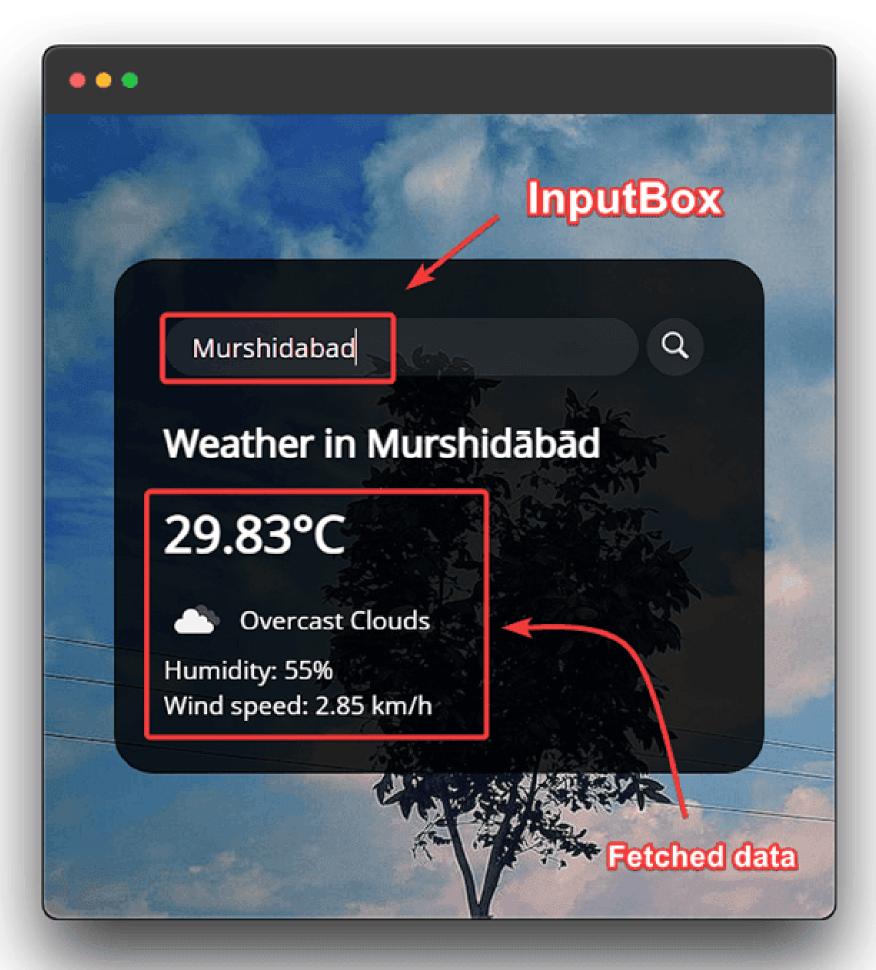
Introduction

App Overview

Visualizing Data: Once you've retrieved the weather data, you need to display it on your web page. This slide will cover how to use CSS and JavaScript to create a visually appealing weather display.



Q



Adding User Interactivity

User Experience: A good weather web app should allow users to interact with the data. This slide will cover how to add user interactivity using JavaScript, such as allowing users to search for specific locations or toggle between different views.

Technologies Used

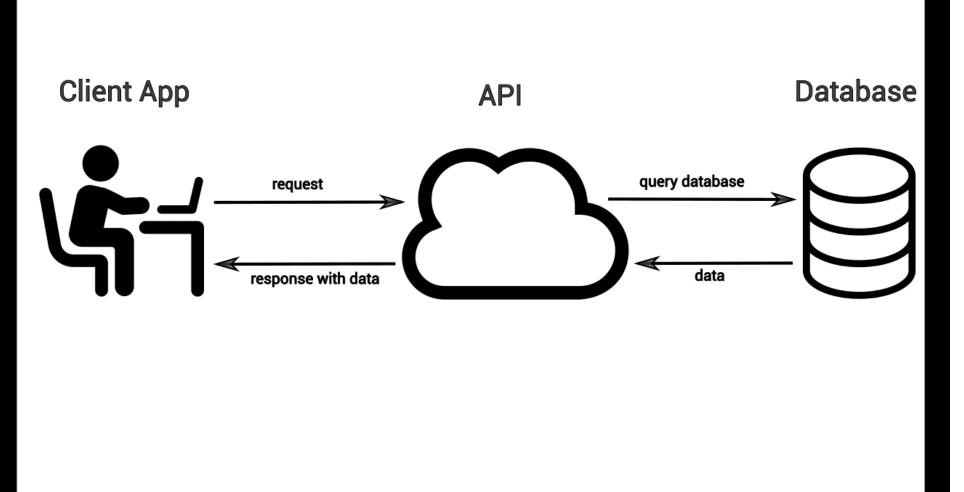
HTML, CSS, JavaScript Our app is crafted using the fundamental trio of web development: HTML for structure, CSS for style, and JavaScript for interactivity. This combination enables a dynamic and visually engaging user experience.Responsive DesignThe app's layout adapts flawlessly to various devices, ensuring user-friendliness across screens.. - -

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Making API Calls:

Real-Time Weather DataOur app sources real-time weather data from the OpenWeatherMap API, a comprehensive weather information provider.

Features:

Current cond information.

Forecast: Accurate predictions for the upcoming days.

Location-based search: Customized weather data for userdefined locations.

API Integration:

We seamlessly integrate the API into our app's JavaScript code, fetching data and dynamically updating the interface.

With the OpenWeatherMap API, our app ensures users are informed and prepared for the elements.

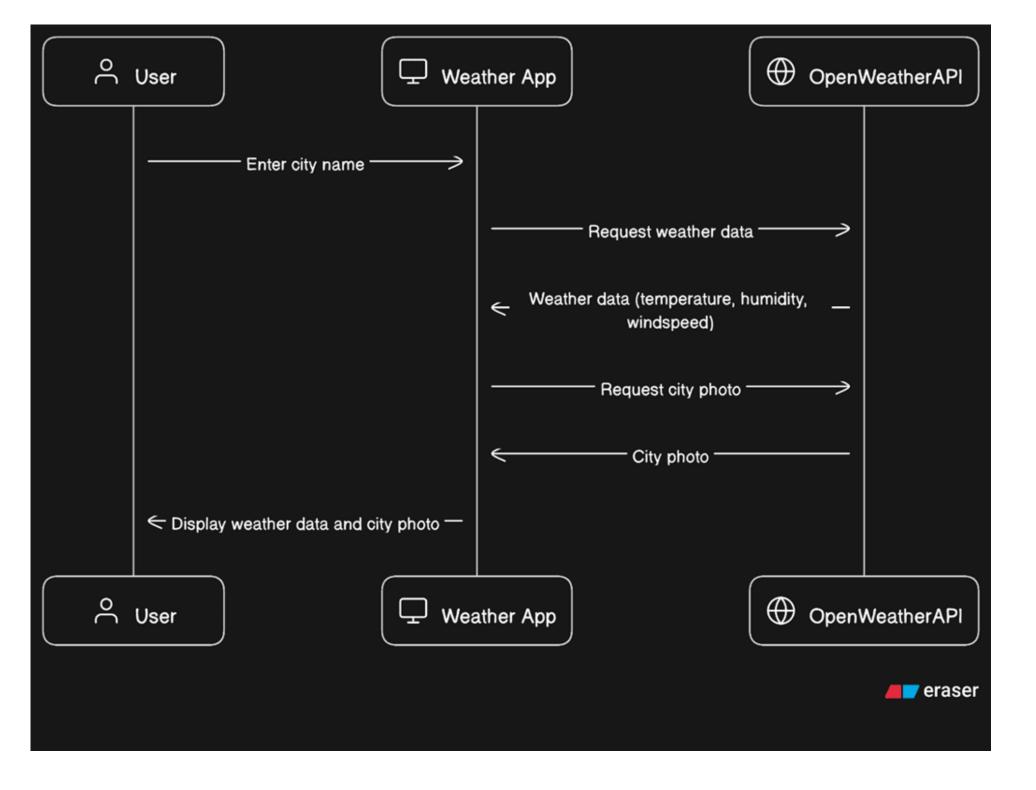
OpenWeatherMap API

Current conditions: Display of up-to-date weather

Data Flow Diagram (DFD)

This represents the main functionalities of the weather app. It consists of two main processes: "Fetch Weather Data" and "Display Weather Data."

- •

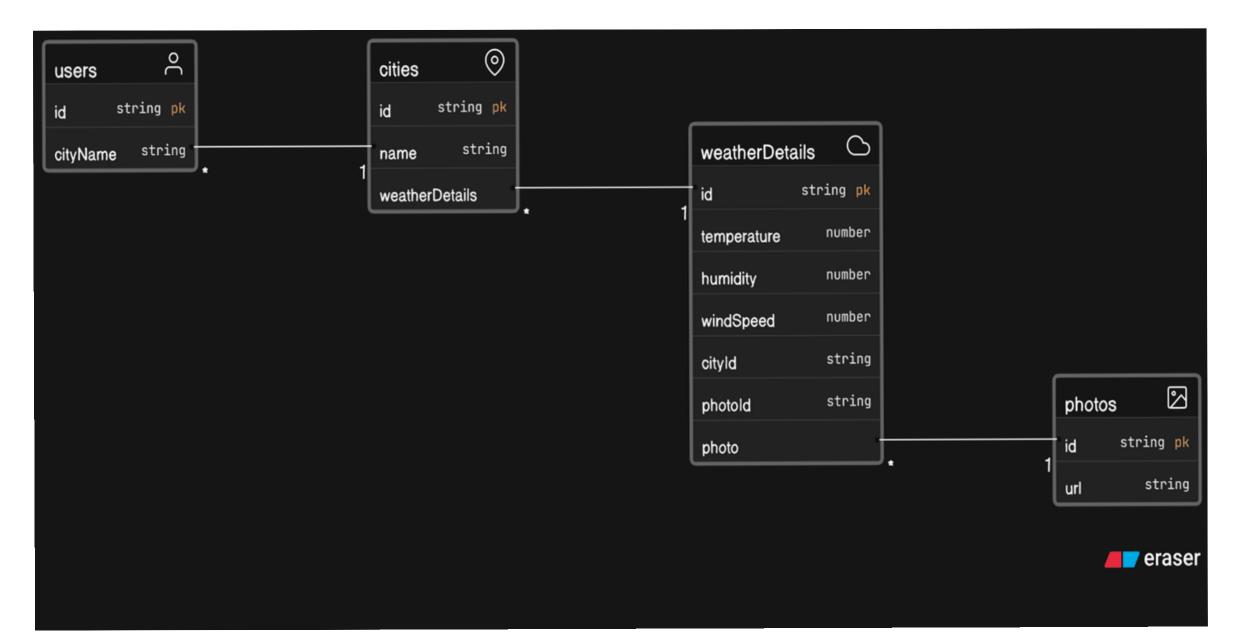




• Fetch Weather Data: This level breaks down the "Fetch Weather Data" process into two subprocesses. The first sub-process, "Send API Request," sends a request to the OpenWeatherMap API to fetch weather data. The second sub-process, "Receive API Response," receives the response containing weather information.

Display Weather Data: This level elaborates on the "Display Weather Data" process. It includes two subprocesses: "Render Current Weather" and "Render Weather Forecast." These sub-processes handle the rendering of current weather information and weather forecasts, respectively.

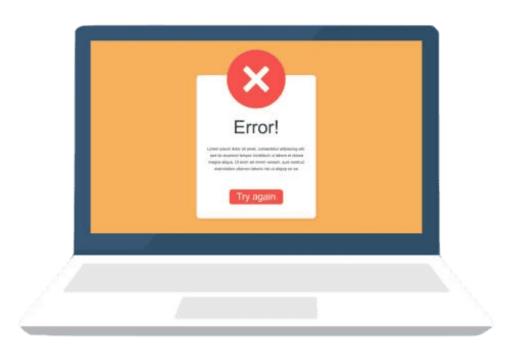
Entity Relationship Diagram (ERD):



• Relationship: The relationship between Location and WeatherData is established through the LocationID foreign key in the WeatherData table, connecting each set of weather data to its corresponding location.

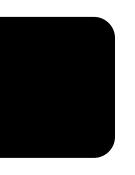
Entities:

- Location: Represents a specific location for which weather data is being fetched. Each location has a unique LocationID and includes attributes like City and Country.
- WeatherData: Stores the weather information retrieved for a particular location. Each set of weather data has a unique WeatherDataID and is associated with a specific LocationID. The attributes Humidity, include Temperature, WindSpeed, and WeatherDescription.



Challenges Faced

- API Integration Complexity: Integrating the OpenWeatherMap API seamlessly into our formats, and handling responses.
- Data Presentation: Effectively visualizing complex weather data in a user-friendly comprehensive information and avoiding overwhelming the user.
- **Responsive Design:** Ensuring that our app looked and functioned well across various usability was a task that required meticulous testing.
- Error Handling: Managing errors that could arise from API calls or unexpected user disrupting the user experience was a delicate balancing act.
- Performance Optimization: Striving for optimal performance, especially when fetching minimizing network requests.



app posed difficulties, requiring thorough understanding of API endpoints, request

manner was a challenge. We needed to strike a balance between providing

devices was a challenge. Adapting the layout to different screen sizes while maintaining

inputs was essential. Implementing robust error-handling mechanisms without

data and rendering it dynamically, required careful consideration of code efficiency and

Conclusion

In conclusion, our Weather App powered by the OpenWeatherMap API represents the synergy of technology and user-centric design. With HTML, CSS, and JavaScript as our tools, we've crafted an application that not only delivers real-time weather data but also offers an intuitive and engaging experience.

Through challenges faced and solutions devised, we've honed our development skills and gained valuable insights into API integration, user interface design, and responsive development. The journey has affirmed the importance of adaptability, collaboration, and continuous learning in the ever-evolving landscape of app development.

As we embrace future enhancements and possibilities, we remain committed to providing users with accurate weather information while ensuring a seamless, visually appealing, and user-friendly interaction. Thank you for joining us on this journey of innovation and exploration.



Acknowledgement

We would like to extend our heartfelt gratitude to all those who contributed to the realization of our Weather App using the OpenWeatherMap API. This project has been a collaborative effort, and we wish to express our appreciation to:

Our teammates, whose dedication and teamwork brought diverse perspectives and creative solutions to the table.

Our teacher, MR. SADEKUL ISLAM whose guidance, insights, and expertise played a crucial role in shaping our understanding of web development and API integration.

Your support has been invaluable in helping us navigate challenges, discover new horizons, and create an app that we're proud to present. Thank you for being an integral part of this endeavor.





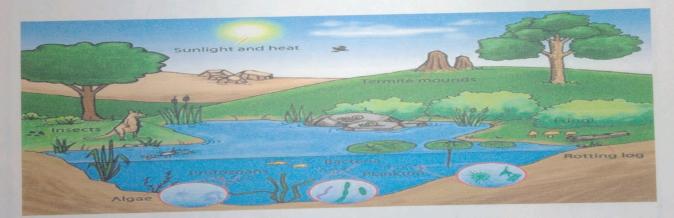


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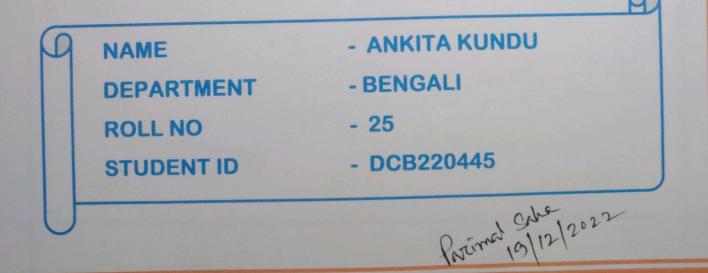
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ENVIRONMENTAL STUDIES

PROJECT TOPIC : STUDY OF RIVER ECOLOGY



B.A (HONOURS COURSE), SEMESTER : 1ST, 2022



ACKNOWLEDGEMENT

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Signature Ankita Kurdu

CERTIFICATE OF GUIDE

This is to certify that Ankita Kundu Student of Dumkal college has successfully completed that project work titled " Study of River Ecology" This project work is the record of authentic work corried out by him/her. she has worked lender my Guidance

Date: - 13/12/2022

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Page no

- 1. Introduction
- 2. Problems
- 3. Objectives.
- 4. Location of study Area.
- 5. Methods of Data collection.

CONTENT

- 6. Results.
- 7. Discussion
- 8. Suggetion
- 9. Limitation of the study
- 10. Référence

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